



Prof Lucimara Stolz Roman

Group coordinator : Nanostructured devices laboratory

Physics Department- UFPR

Vice Coordinator: Central Laboratory of Nanotechnology (SiSnano)



Clean Energy and Sustainability Symposium: Australia-Brazil – Organic Solar Cells



Australian Government
Department of Education



Federal University of Parana



12 Academic Units with 70 Departments

- Agronomic Sciences
- Applied Social Sciences
- Architecture/Urb & Engineering (Technology)
- Biological Sciences
- Earth Sciences
- Education
- Exact Sciences
- Health Sciences
- Human Sciences, Literature and Linguistics
- Law School
- Oceanic Sciences Research Center
- Professional and Technological Education

1912 - Foundation of the UFPR

1967 - First **Master/Doctoral** Program

2004 - the **Patent Office**

2008 - the **Technology Innovation Office**



107 Undergraduate courses

105 Specialization courses

90 Master / PhD Programs

26,000 undergraduate students

5,500 graduate students

2200 faculty members

3450 staff

Total land: 9.043.913 m² **Buildings:** 395.792 m²



About 1900 Master/PhD titled per year





Forestry Technology and Engineering Department - UFPR
 Wood Quality and Anatomy Laboratory (LANAQM)
 Agroforestry Nanotechnology Group (GNanoAgro)

Prof^ª. Graciela Ines Bolzon de Muniz
 (gmunize@ufpr.br),

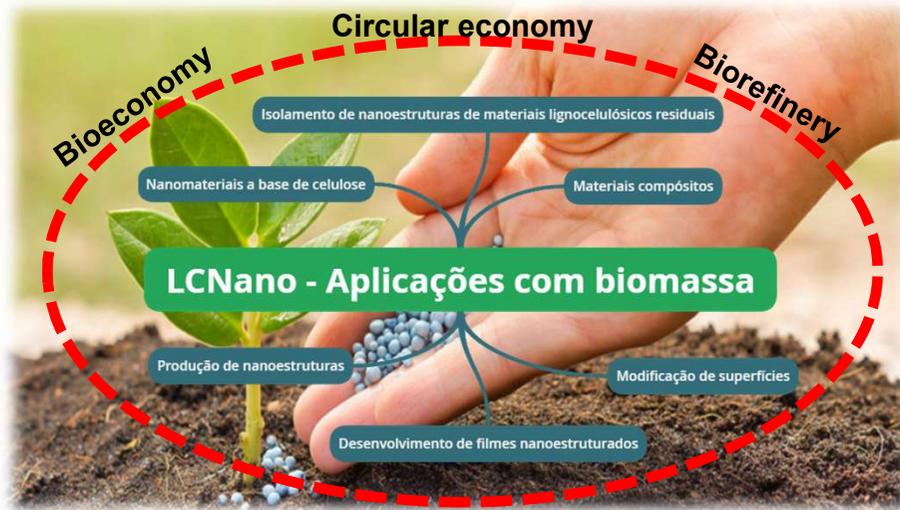
Prof^º Pedro H. G. de Cademartori
 (pedroc@ufpr.br / www.linkedin.com/in/phgcademartori),

and a multidisciplinary team with more than 20 researchers.

Partnerships with industries, institutes and research networks for prospection of new products based on nanotechnology, biorefinery, circular economy and bioeconomy concepts.

Main research interests

- New products and applications for nanocellulose;
 - New products and applications for lignin;
- New products and applications for fast-pyrolysis bio-oil and its derivatives;
 - Development of nanocomposites;
 - Application of nanostructured biocides;
- Surface modification of lignocellulosic materials.





Luiz Pereira Ramos

Department of Chemistry
Federal University of Paraná (UFPR)
Curitiba, PR, Brazil

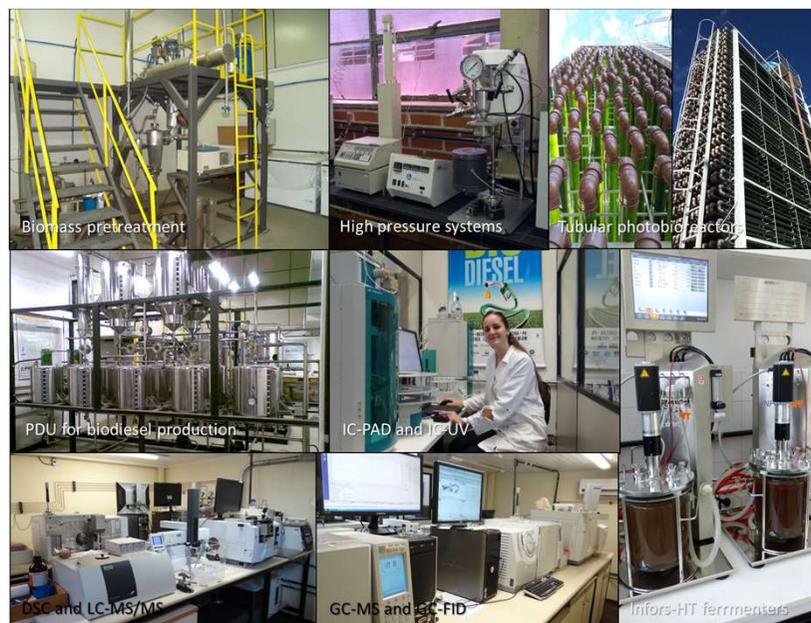
www.quimica.ufpr.br/paginas/luiz-ramos

luiz.ramos@ufpr.br



Research Center in Applied Chemistry

- I. Carbohydrate and lignin chemistry
- II. Heterogeneous catalysis for biodiesel production
- III. Enzyme-mediated biofuel processes
- IV. Microalgae conversion to fuels and chemicals
- V. Sugar platform – second generation carbohydrates and carbohydrate derivatives
- VI. Multifunctional materials – natural and synthetic layered materials, bionanocomposites (nanocellulose)
- VII. Process intensification (SCF, microwave, ultrasound)
- VIII. Glycerin and lipid chemistry – plasticizers, polymers, precursors and fine chemicals

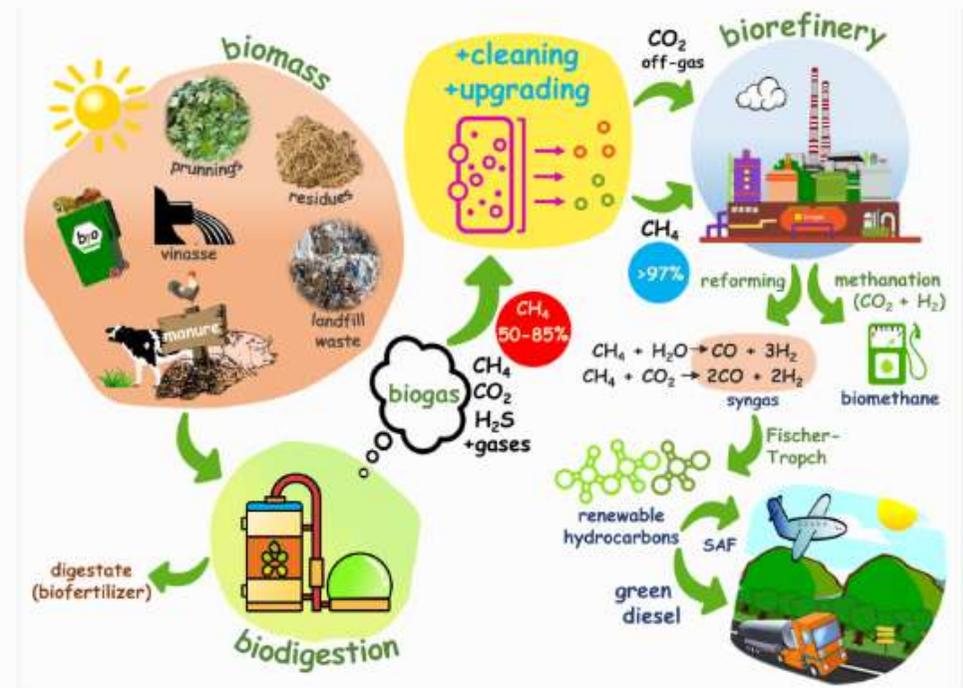


PtX Paraná Project - Energy Transition & Low Carbon Economy in the Paraná's Agribusiness

Objective

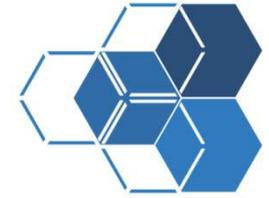
To enable a new technological route to produce synthetic sustainable fuels from biogas through the integration of catalytic reforming and Fischer-Tropsch synthesis, with emphasis on developing the Power-to-X market in the State of Paraná.

Project partners



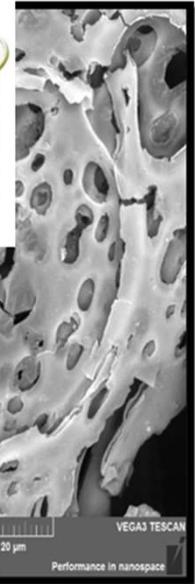
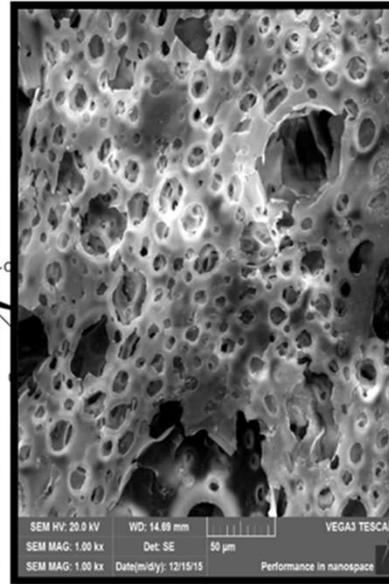
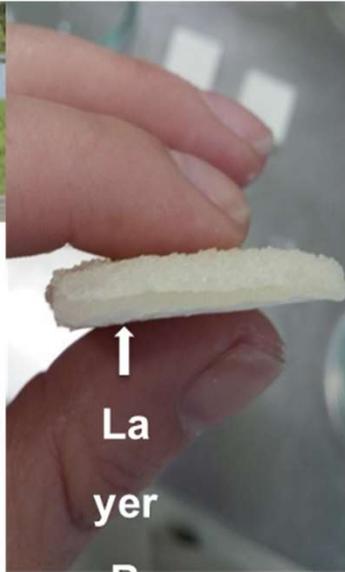


- **Prof. Helton José Alves**
- Dean of Research at Federal University of Parana



LCNano

Obtaining nanochitosan on a pilot scale and developing products with greater added value.



LABCATPROBIO
LABORATÓRIO DE CATÁLISE E
PRODUÇÃO DE BIOCOMBUSTÍVEIS

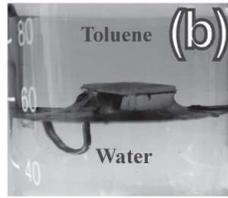
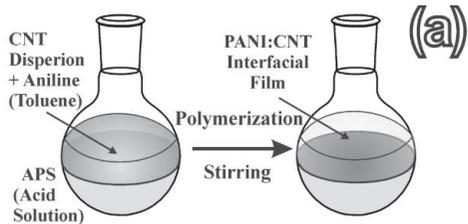


Carbon nanotubes- and graphene-based nanocomposites: preparation, characterization, applications (photovoltaics, sensors, batteries, etc.)

Aldo Zarbin Chemistry department



LCNano



(c) These studies, the effect of the conductivity of polyaniline on the redox reactions and the application of polyaniline films are as follows: (i) Polyaniline is oxidized by a transfer of electrons from the diamine structure to a so-called proton addition, which increases the conductivity of the polyaniline structure. The protonation reactions by dispersion

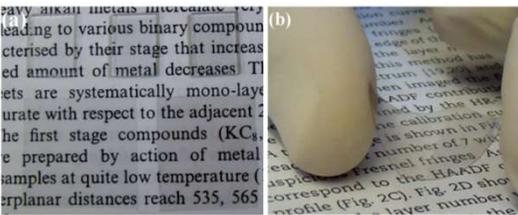
Thin, transparent, conductive films deposited over ordinary substrates

can be conveniently studied by Raman spectroelectrochemistry. From the experimental results of this research, it seems clear that in protonated polyemeraldine (the only conducting form of polyaniline) structurally nonequivalent segments coexist. The

CNTs/polyaniline

Several studies have been conducted in recent years on the synthesis, characterization and applications of CNT/polyaniline nanocomposites. A dramatic improvement in mechanical, thermal, electrical, optical, and redox properties

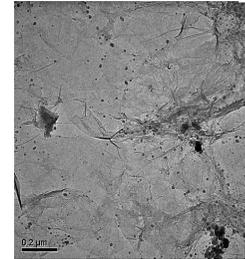
graphene/polyaniline



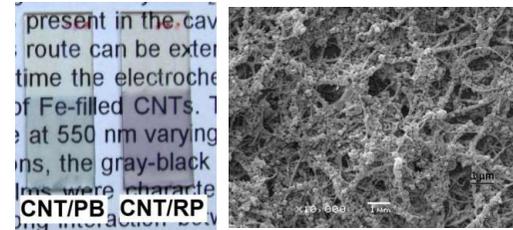
Graphene and tri-layer graphene

extended X-ray absorption fine structure analysis of the precursor and the thiol, resulting in the formation of a Pt hydride. The Pt nanoparticles were formed by the introduction of dodecanethiol. The thiol was used to control the size of the Pt nanoparticles with fcc structure and to transform infrared spectroscopy

CNT/polythiophene



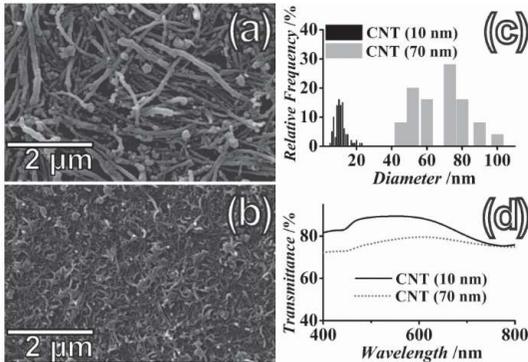
Graphene/Ag nanoparticles – SERS substrates



CNT/Prussian blue: sensor; electrochromic device; electrocatalyst



CNT- and graphene/natural Rubber/ structural and multifunctional material



Grupo de Química de Materiais

Organic Solar Cells

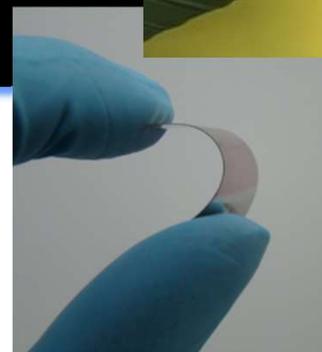
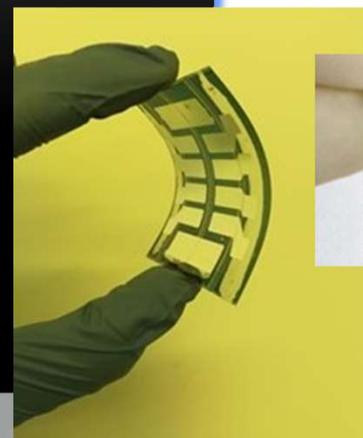
Motivation

Large area devices

Easy processing

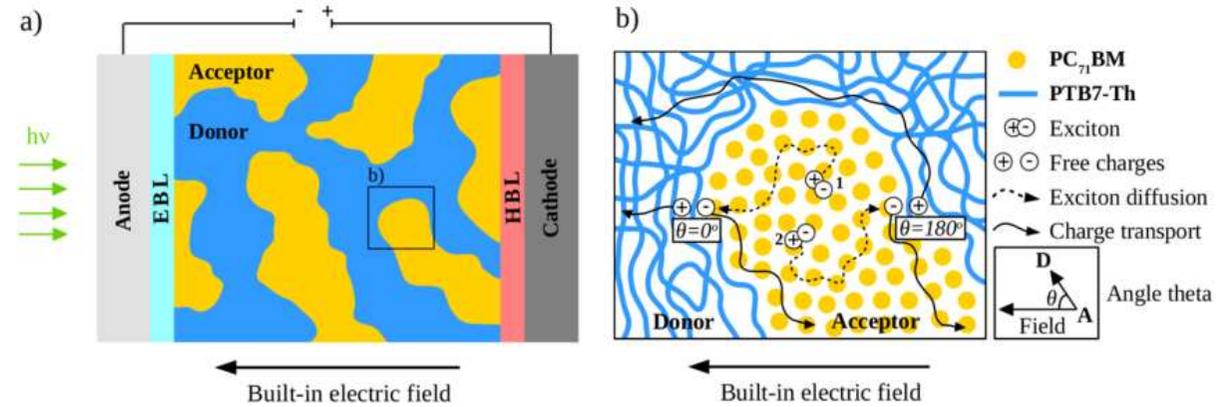
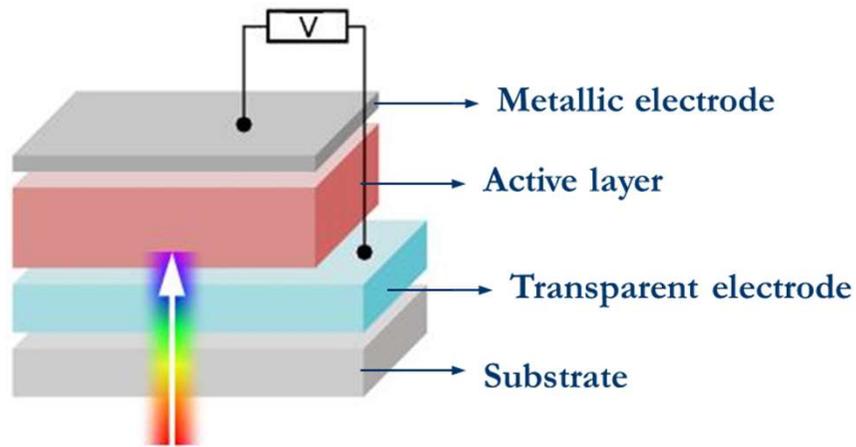
Possibly low-cost devices

Flexible devices



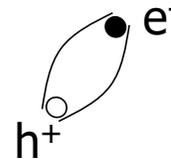
The devices are constructed in a sandwich geometry.

The electrodes must have different work function value to create a built-in potential inside the active layer helping the charge collection.



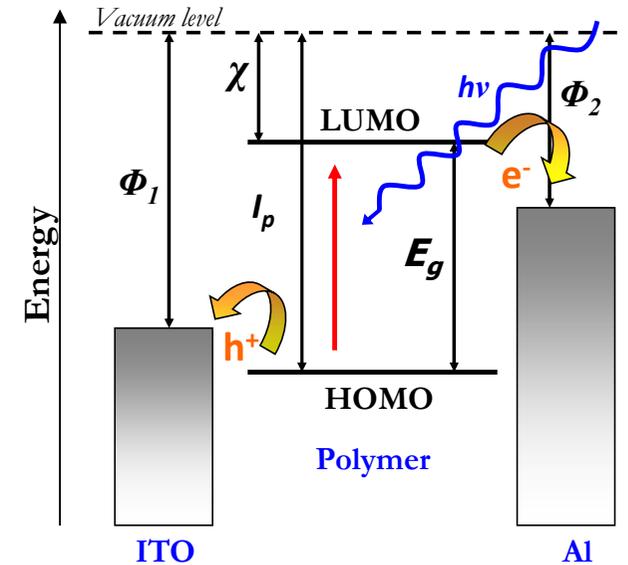
$$E = \frac{e^2}{4\pi\epsilon_0\epsilon r}$$

exciton

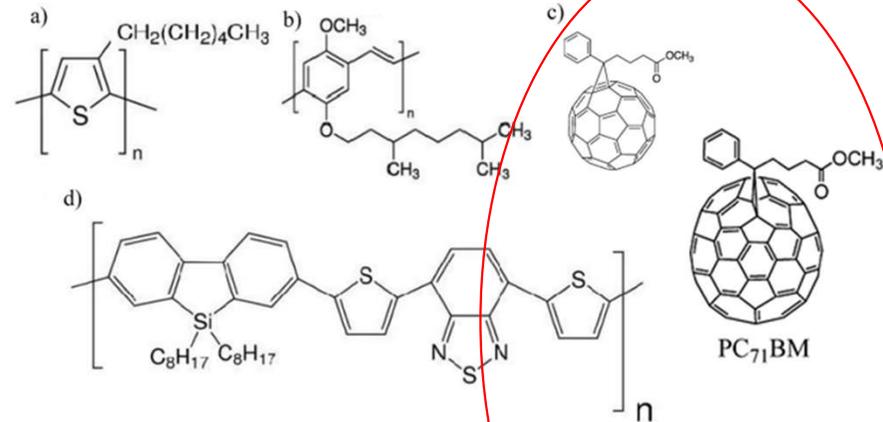
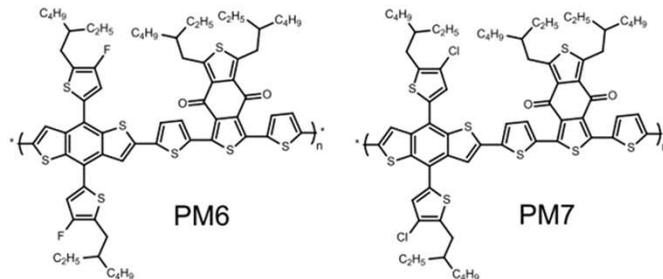


The mechanism of photocurrent generation include basically four processes:

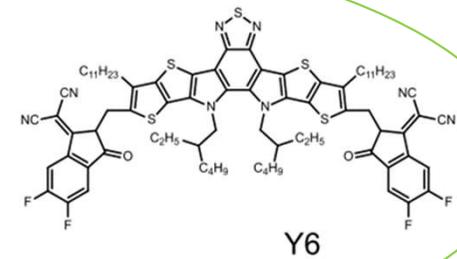
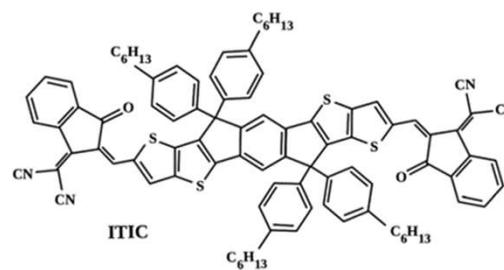
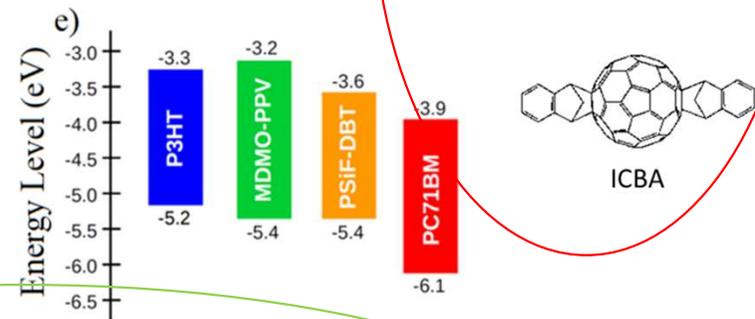
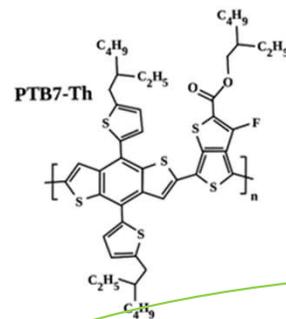
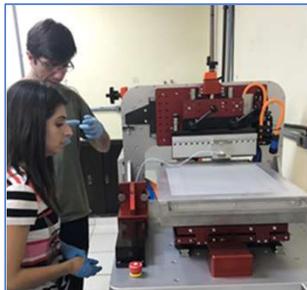
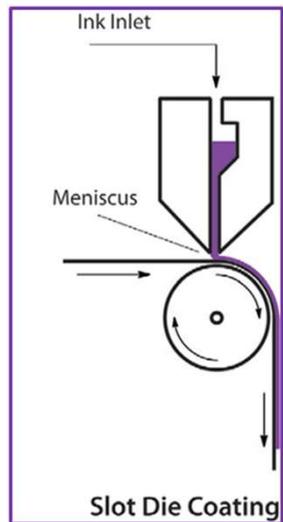
- i) light absorption and generation of excited states;
- ii) the diffusion of excited states to sites where dissociation of excitons may occur;
- iii) dissociation of the excitons to form free charge carriers (at donor acceptor interfaces);
- iv) and transport of the carriers by drift and diffusion to the respective electrode for collection.



Examples:



Spin coating deposition



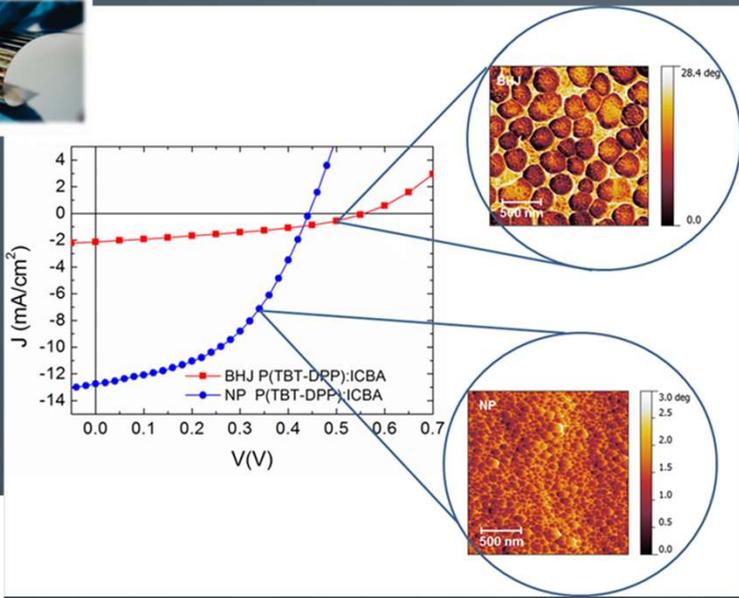
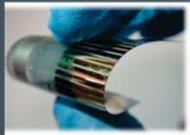
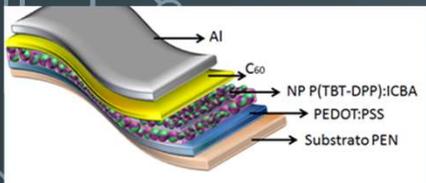
Outline



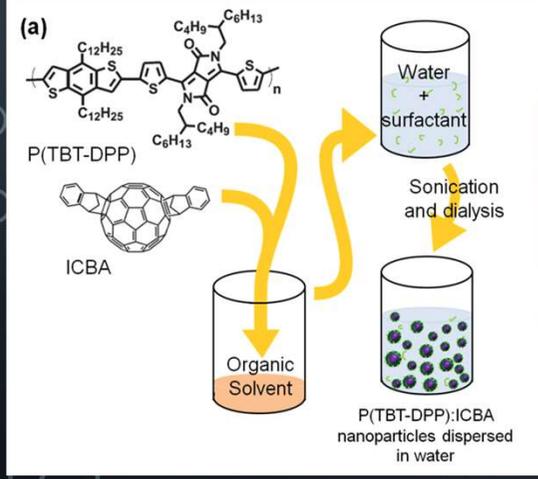
ORGANIC SOLAR CELLS (OSCs):



- (i) OSCs – Green solvent; Ternary blends, FRET, NPs on water;**
- (ii) OSCs – improving stability by doping D/A blends;**
- (iii) OSCs - Studies the case on OSCs lifetime.**

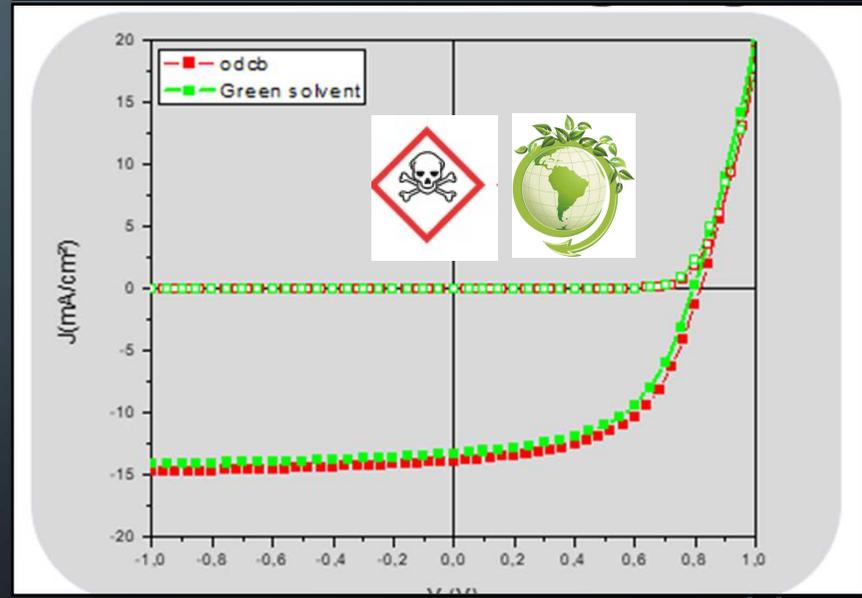
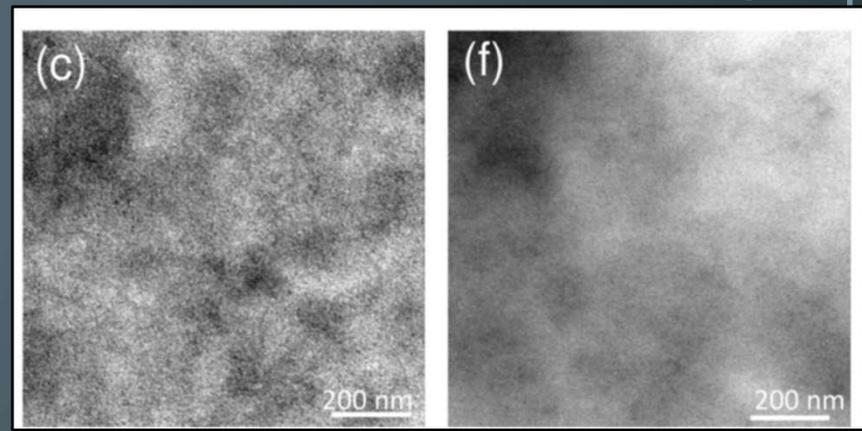


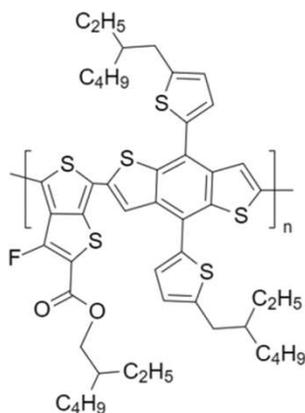
	J_{sc} [mA cm ⁻²]	V_{oc} [V]	FF [%]	PCE [%]
BHJ	2.12	0.55	37	0.44
NP	12.73	0.44	47	2.63



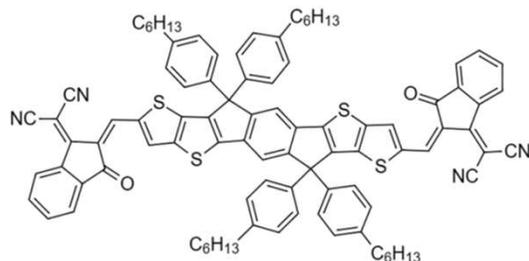
Non Miscible materials

Miscible materials

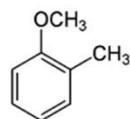




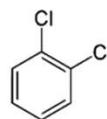
PTB7-Th



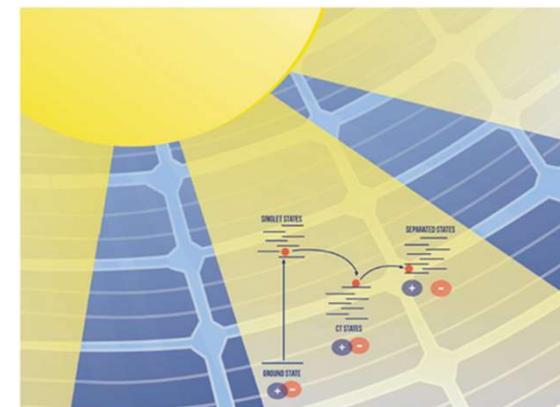
ITIC



o-MA



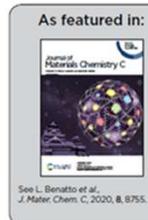
o-DCB



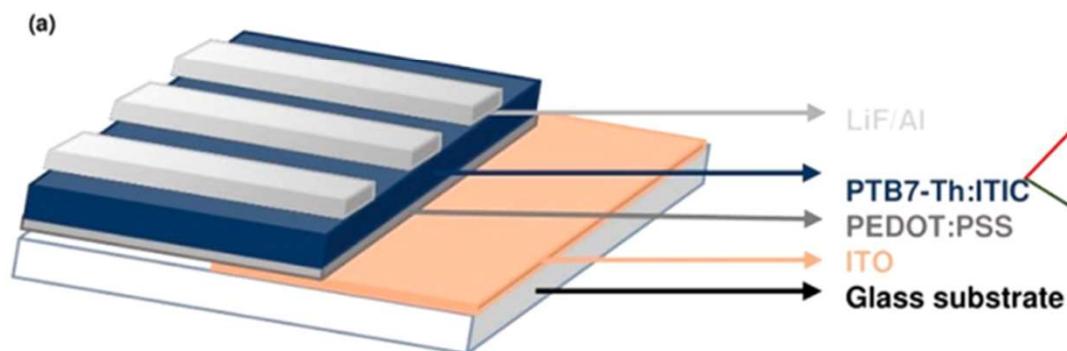
Showcasing collaborative research from Federal University of Paraná, Curitiba, Brazil and Center of Innovations, CSEM Brazil, Belo Horizonte, Brazil

Kinetic model for photoluminescence quenching by selective excitation of D/A blends: implications for charge separation in fullerene and non-fullerene organic solar cells

Our findings demonstrate the importance of kinetic factors to determine the overall charge separation efficiency in donor/acceptor systems for organic photovoltaic applications. Providing understanding of the charge transfer dynamics to further improve the efficiency of organic solar cells.



See L. Benatto et al. *J. Mater. Chem. C*, 2020, 8, 8755.

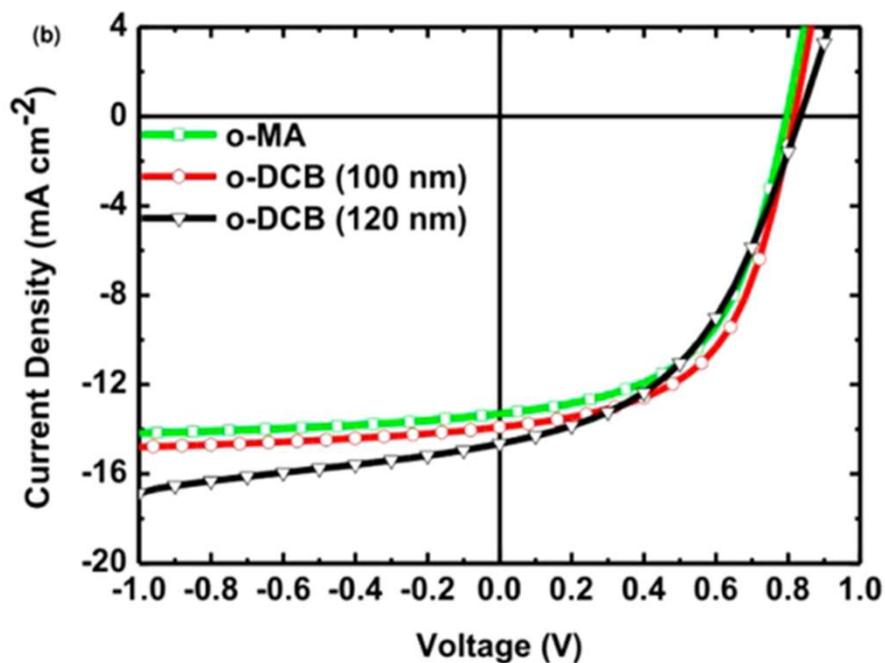


rsc.li/materials-c
Registered charity number 207890

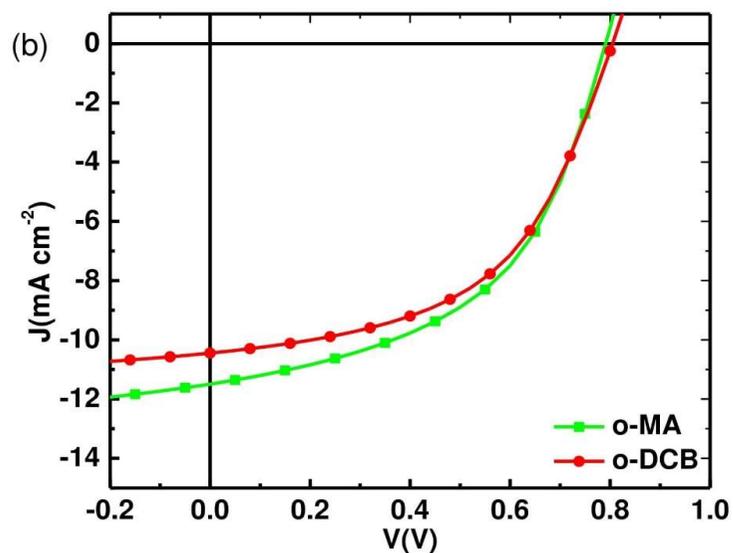
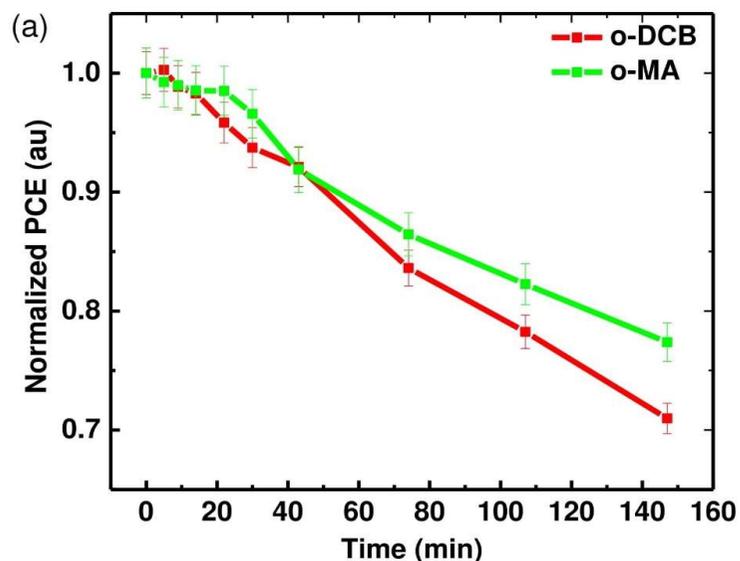
L. Wouk, Y. Jin, L. Benatto, C. Wang, M. Koehler, F. Zhang, LS Roman, *ACS Appl. Energy Mater.*, **2018**, 1 (9), pp 4776–4785

Table 1. Photovoltaic Parameters of the OPVs Based on PTB7-Th:ITIC BHJ, with D:A Ratio of 1:1.3 (Average Values of 20 Devices Are Provided in Parentheses)

devices	thickness (nm)	J_{sc} (mA cm ⁻²)		V_{oc} (V)		FF (%)		PCE (%)	
<i>o</i> -MA	120 ± 5	13.31	13.40 ± 0.40	0.79	0.79 ± 0.01	53.7	52.2 ± 1.0	5.69	5.36 ± 0.30
<i>o</i> -DCB	100 ± 5	13.91	13.71 ± 1.00	0.82	0.82 ± 0.01	54.7	53.7 ± 0.5	6.21	5.90 ± 0.20
<i>o</i> -DCB	120 ± 5	14.65	14.0 ± 1.00	0.83	0.83 ± 0.01	45.7	44.3 ± 1.0	5.58	5.12 ± 0.30

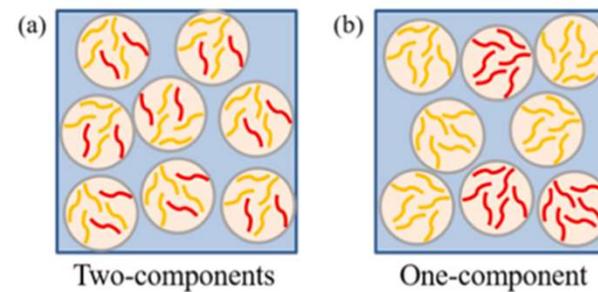
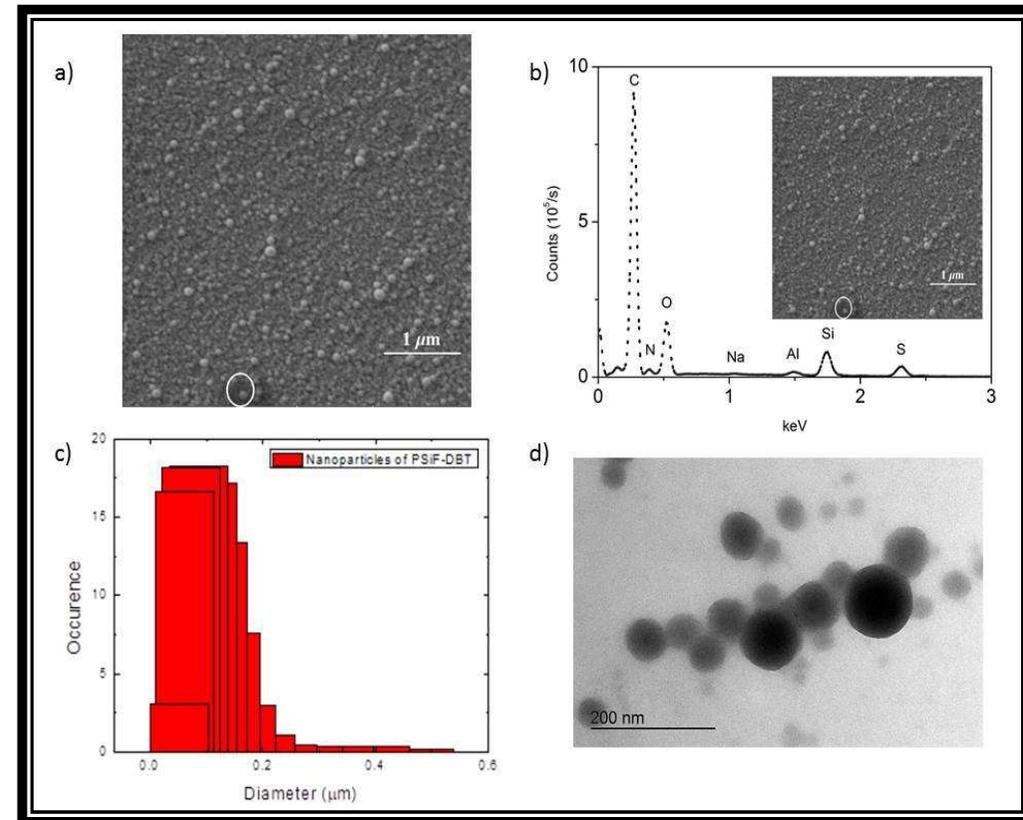
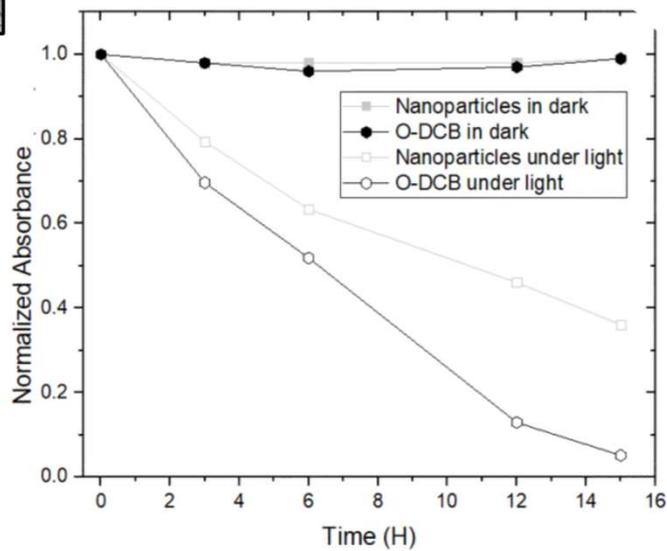
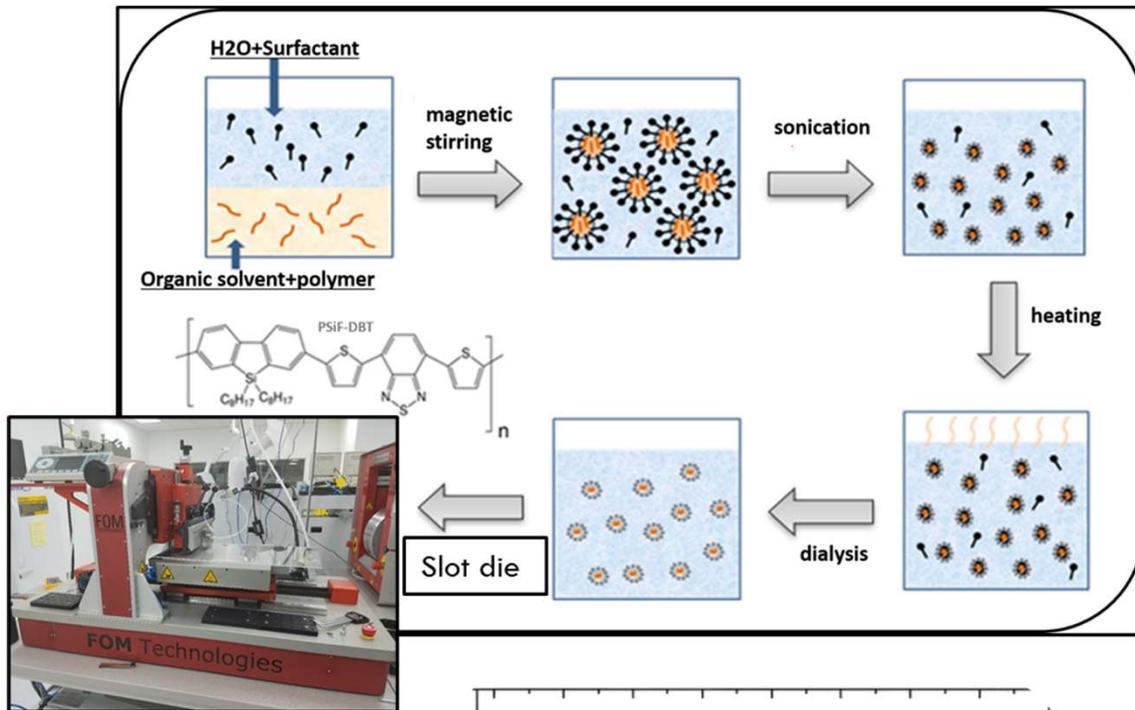


o-Ma (120nm) : Hole mobility = 1.3×10^{-4} cm V⁻¹ s⁻¹
 Electron mobility = 8.0×10^{-5} cm V⁻¹ s⁻¹
o-DBC (100nm) : Hole mobility = 7.9×10^{-4} cm V⁻¹ s⁻¹
 Electron mobility = 2.4×10^{-5} cmV⁻¹ s⁻¹



Stability evaluation of the air-tested and air-processed OSC performance composed by PTB7-Th:ITIC (1:1.3) in o-DCB and o-MA. The air-tested devices (a) were made at glovebox and investigated in more than 2 hours in environmental condition. The active layer of the air-processed devices (b) was spin-coated at room condition.

Devices	Thickness	J_{sc}	V_{oc}	FF	PCE
	(nm)	(mA cm ⁻²)			
o-MA	120±5	11.49	10.24±1.00	0.79	0.79±0.01
o-DCB	100±5	10.45	10.01±0.40	0.81	0.80±0.01



<https://nanocalc.org/fret>

FRET–Calc: A free software and web server for Förster Resonance Energy Transfer Calculation

Leandro Benatto^{a,b,*}, Omar Mesquita^a, João L.B. Rosa^b, Lucimara S. Roman^b,
 Marlus Koehler^b, Rodrigo B. Capaz^{a,c}, Graziãni Candiotto^{a,**}

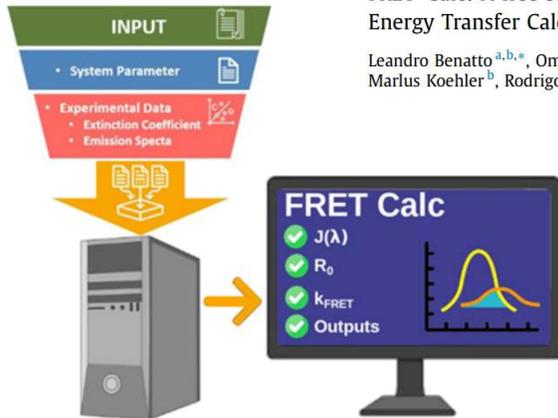
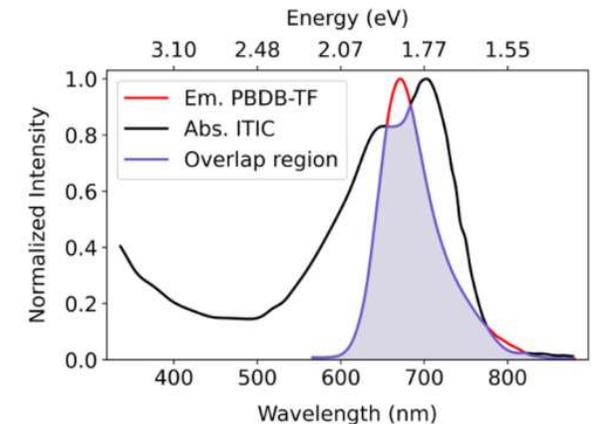
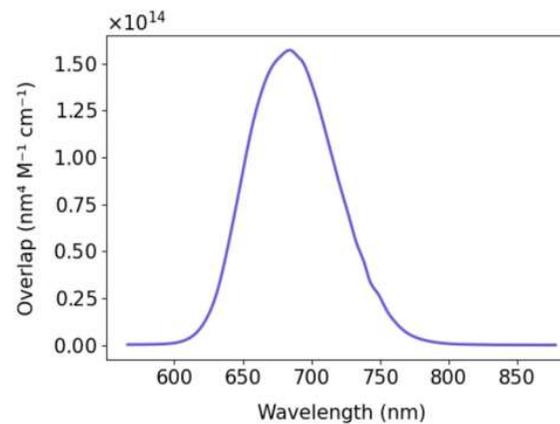
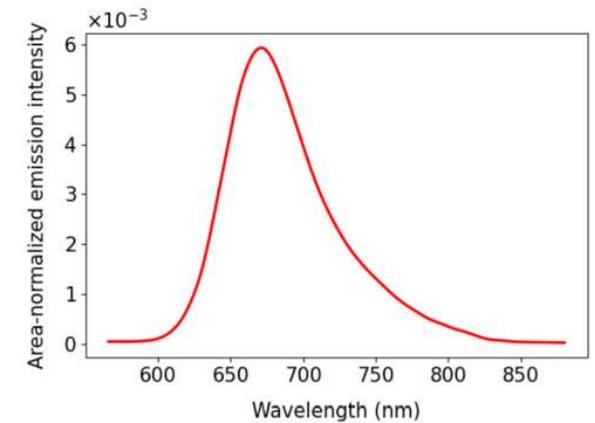
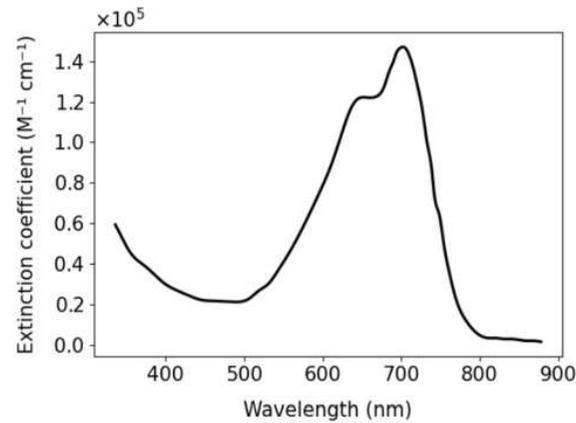


Fig. 1. Graphical representation of FRET–Calc workflow.



Outline

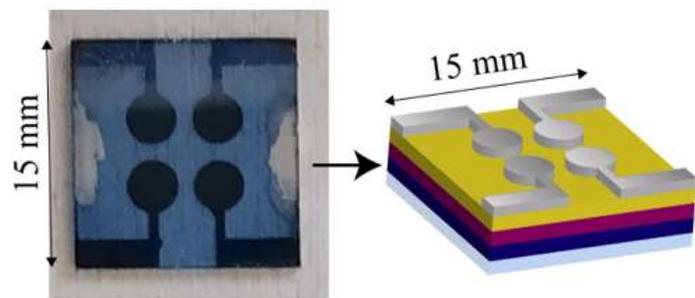


ORGANIC SOLAR CELLS (OSCs):



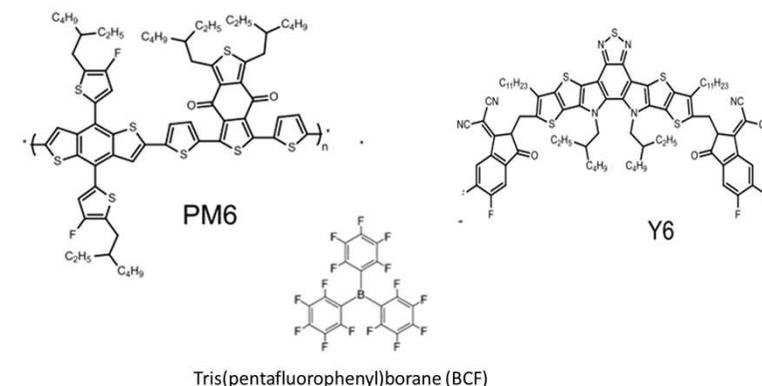
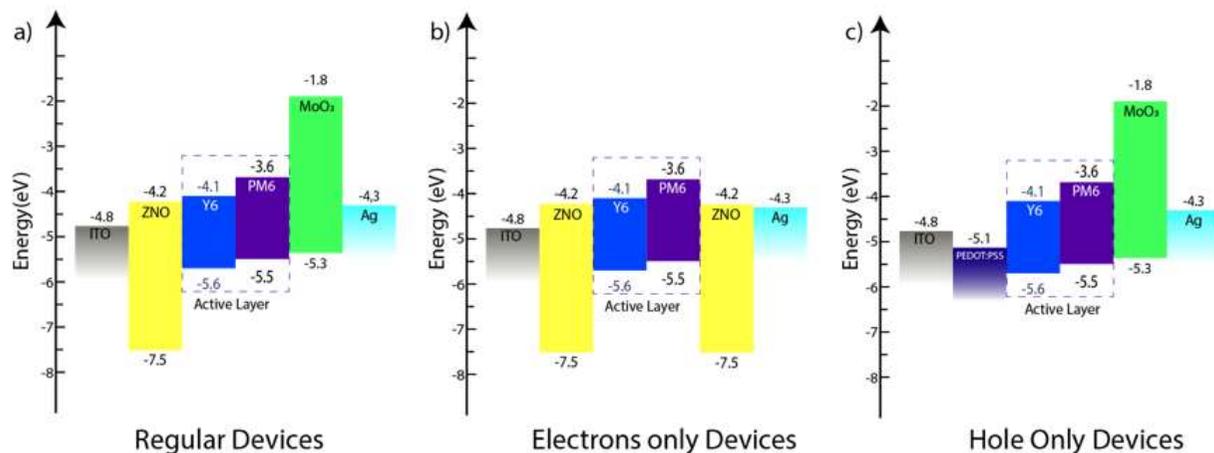
- (i) OSCs – Green solvent; Ternary blends, FRET, NPs on water;
- (ii) OSCs – improving stability by doping D/A blends;**
- (iii) OSCs - Studies the case on OSCs lifetime.

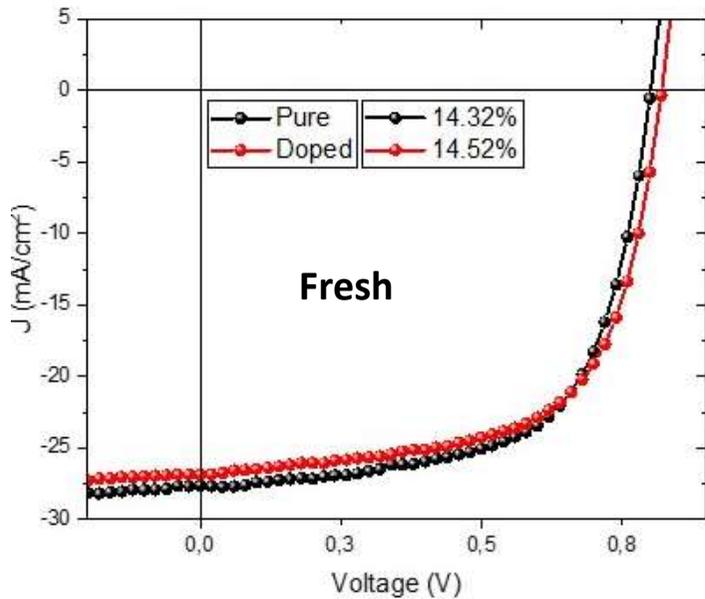
Reduction of OCS degradation through humidity control using BCF in the PM6:Y6 (:1.2) active layer



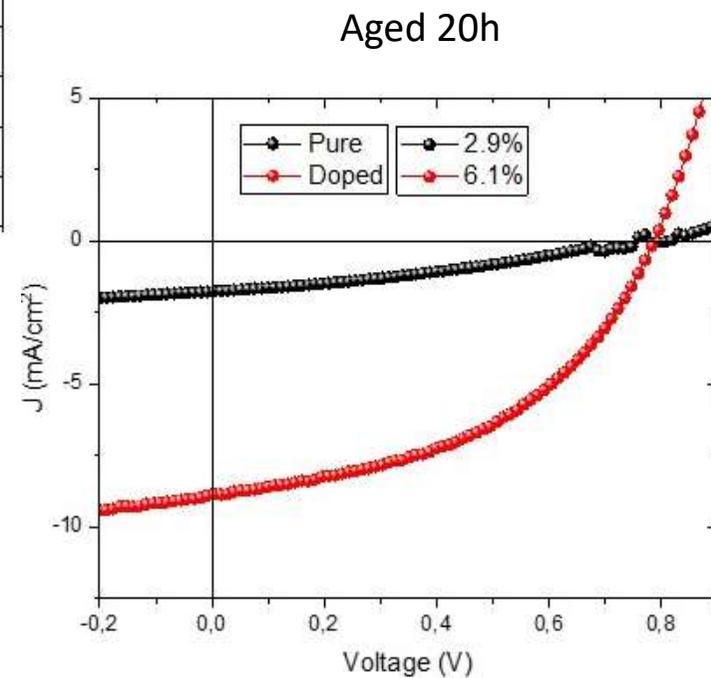
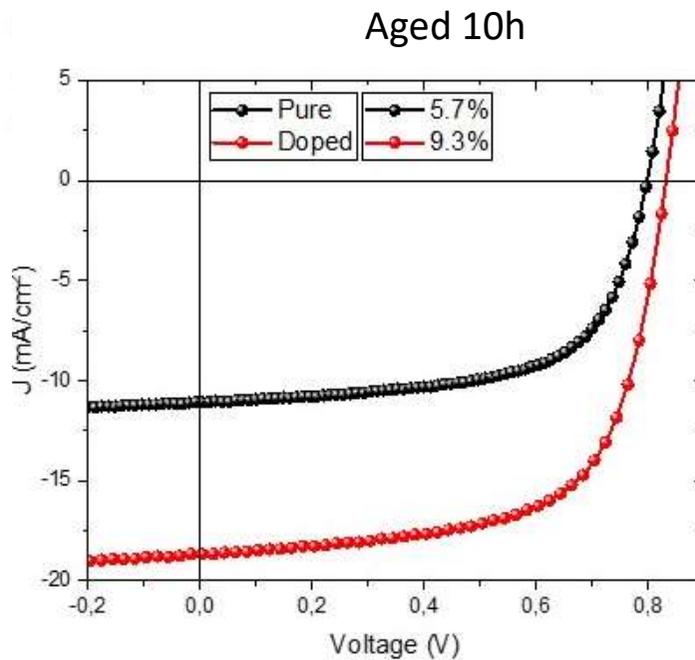
BCF RATE (%)	PCE (%)		
	Fresh	Aged 5h	Aged 10h
0	14.3±0.2	10±0.2	8±0.1
10	2.5±0.1	-	-
5	4.9±0.1	-	-
3	5.6±0.3	-	-
2	14.6±0.1	12±0.1	10±0.2
1	14±0.2	10±0.1	8±0.2

Refiz a tabela do paper das % dos dispositivos com 1 e 2 %





PM6:Y6 with 2% BCF



Outline



ORGANIC SOLAR CELLS (OSCs):
MORPHOLOGY, LIGHT HARVESTING, EXCITON DISSOCIATION AND
ENERGY TRANSFER STUDIES:



- (i) OSCs – Green solvent;
- (ii) OSCs – improving stability by doping D/A blends;
- (iii) OSCs - Ternary blends, FRET;
- (iv) OSCs - Studies the case on OSCs lifetime.**

OPVs for windows:

- Insulfim analysis
- OPVs CSEM/SUNEW on bus station "TUBO".

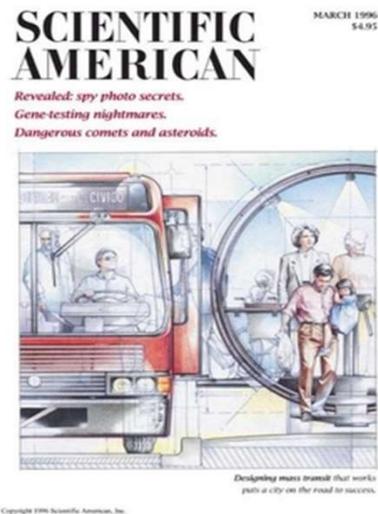






Figura 3 - Painel DPV fixado na folha semirrígida de ACM (imagem ilustrativa).

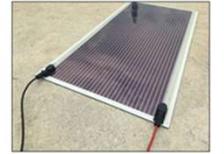


Figura 4 - Painel DPV fixado na folha semirrígida de ACM (imagem real).

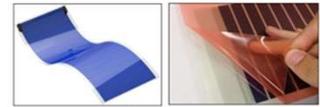
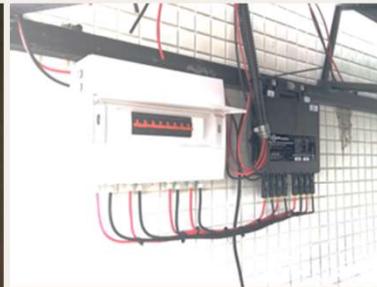
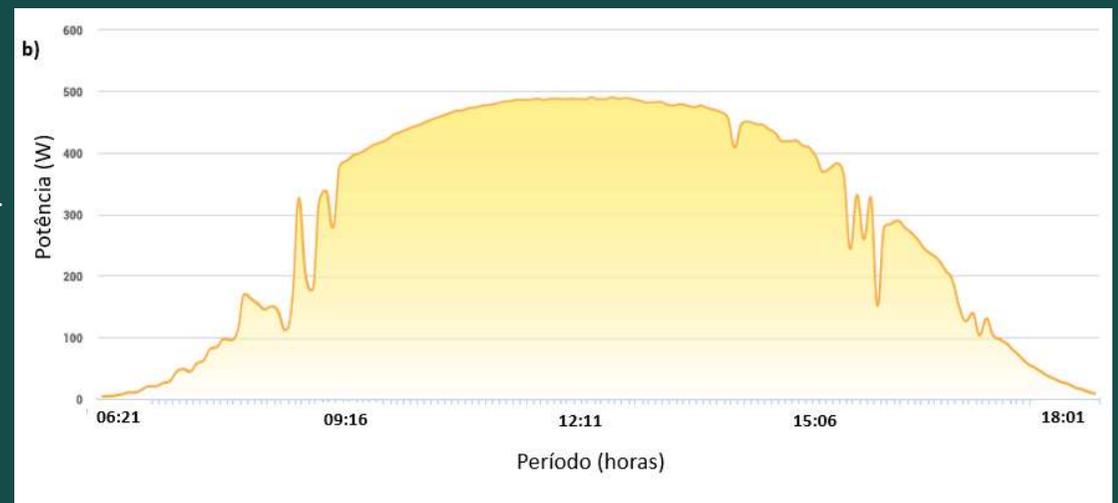
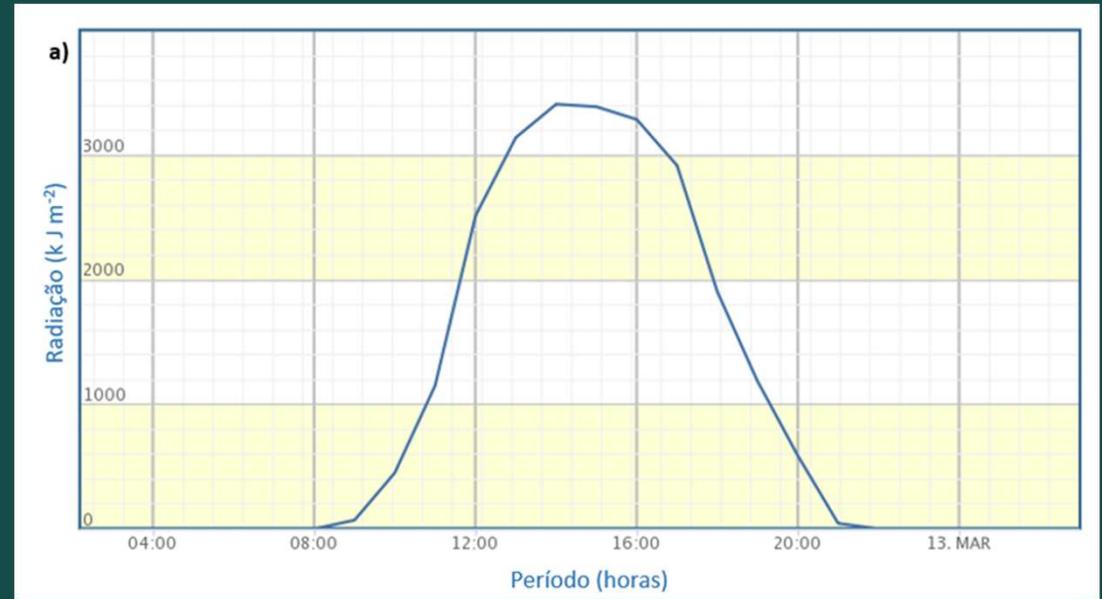
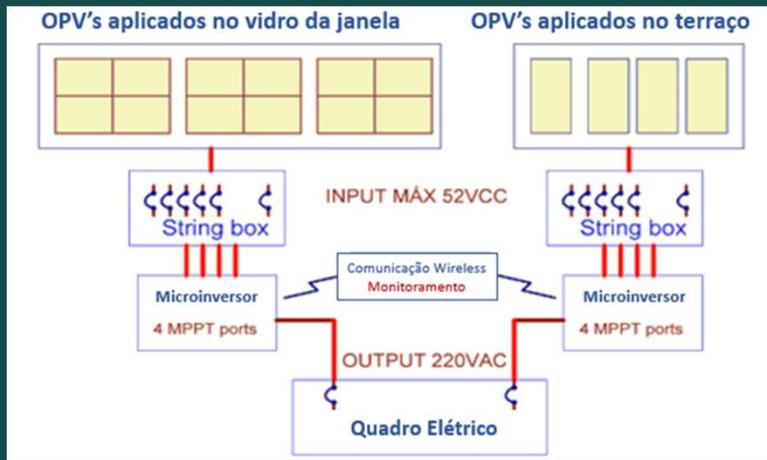


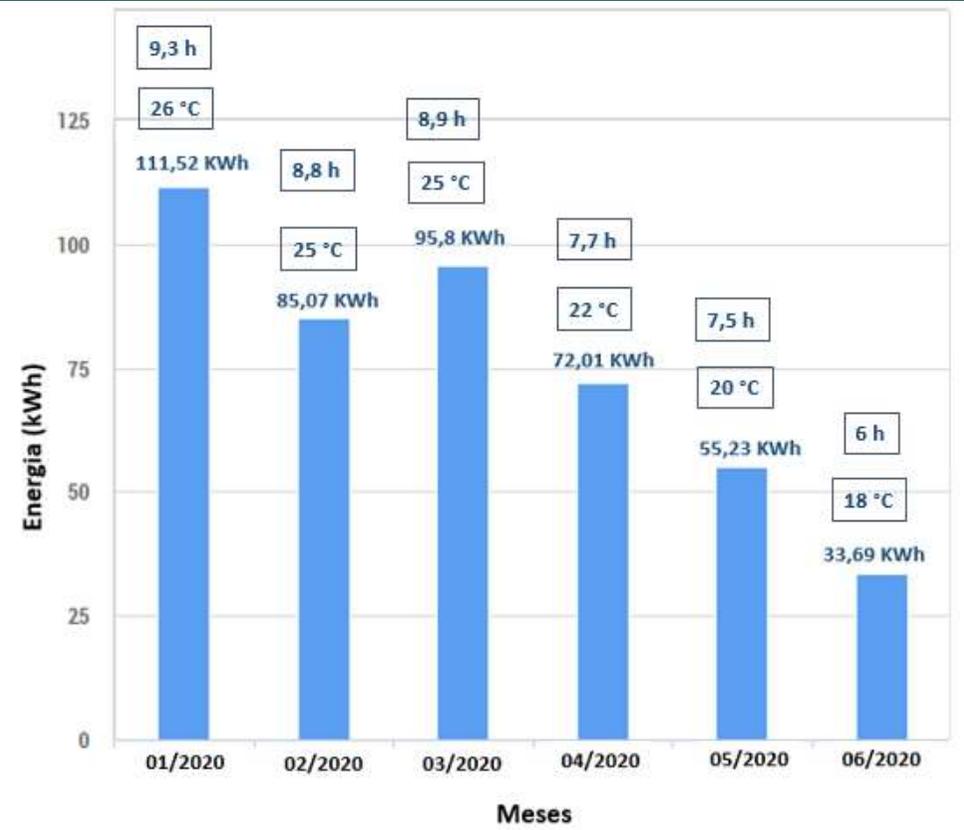
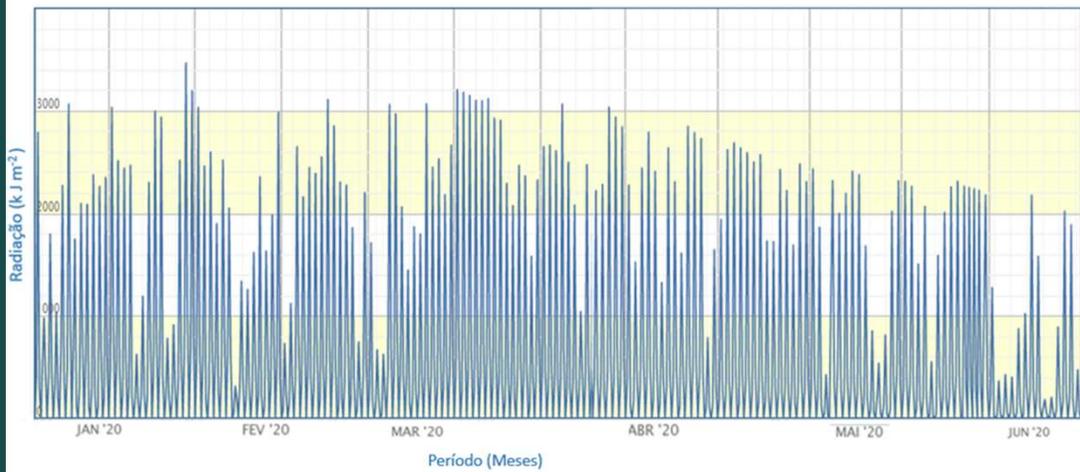
Figura 5 - Instalação do SICHW DPV ILM.

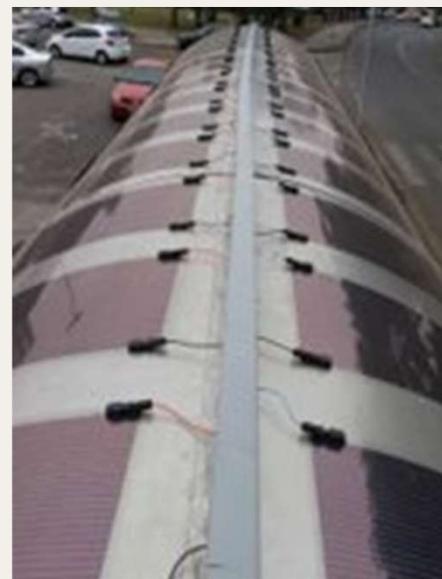




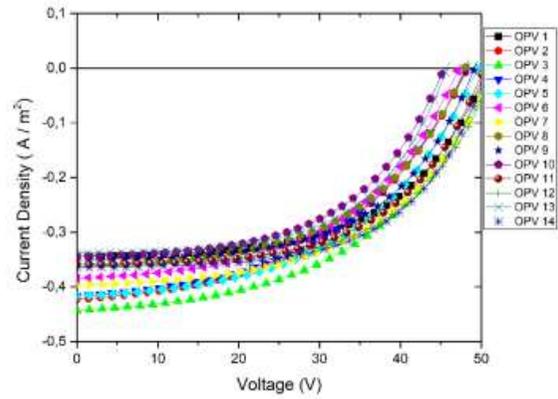
a) Radiação Solar e b) potência solar gerada no dia 12 de março de 2020.
 b) Fonte: INMET e Microinverter Appsystem[®], respectivamente.

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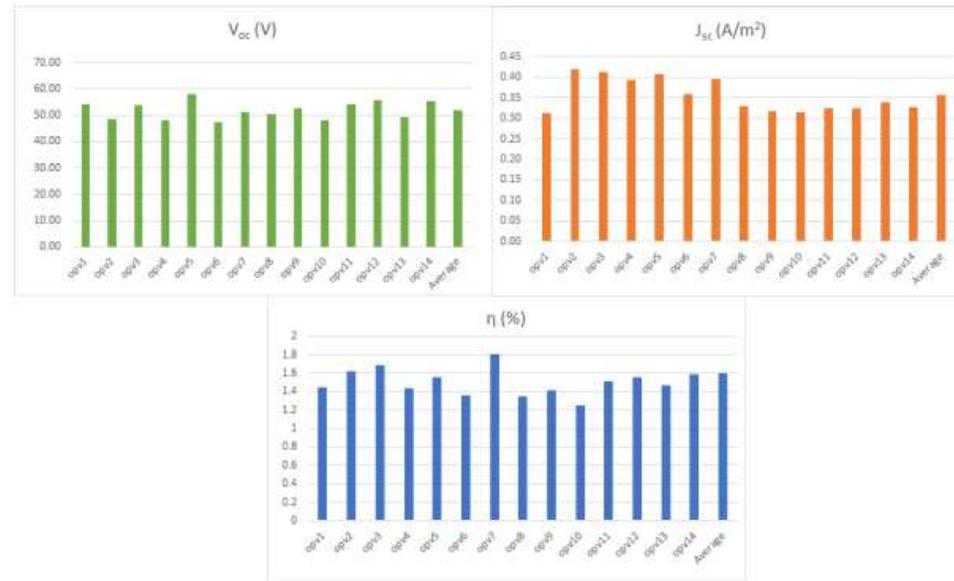




b)



c)





DiNE
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SOBRE ENERGIA SOLAR

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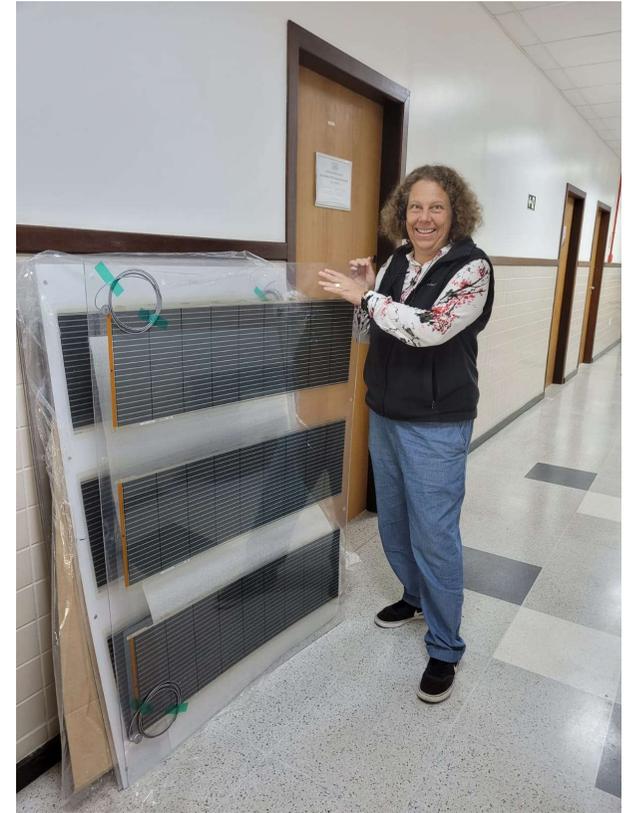
**SOBRE A
ESTAÇÃO TUBO**

- Aplicação do uso de dispositivos fotovoltaicos orgânicos.
- Vantagens da energia solar para o meio ambiente.
- Física dos fotovoltaicos orgânicos.

Also on top of the “TUBE” station and on the roof:



New Project is coming up:
Let`s see how these encapsulated
OPVs work 😊



Acknowledgement:



<https://dineufpr.wixsite.com/dineufpr>

